Olefinic-Amide and Olefinic-Lactam Cyclizations

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ABSTRACT

Olefinic-amide and olefinic-lactam cyclization reactions that result in the generation of cyclic enamides are described.

Small molecules that contain electron-rich olefins (enol ethers and enamides) are both valuable in their own right and are interesting precursors to more elaborate substrates. 1,2 Over the past few years we have been interested in the generation and use of cyclic enol ethers and have recently described their synthesis from olefinic-ester and olefinic-lactone cyclizations using an in situ generated Ti reagent that is presumed to be a Ti ethylidene. Although the corresponding olefinic-amide cyclizations would be synthetically useful, to the best of our knowledge there are only two reports of related reactions that employ amides. Takeda has described Ti(II)-promoted cyclizations of dithianes having pendant amides, and in work more closely related to that proposed

here, Bennasar and co-workers have successfully carried out two-step olefinic-amide cyclizations to enamides. The Bennasar chemistry involves the initial conversion of amides into mixtures of cyclic and acyclic enamides using dimethyltitanocene followed by the conversion of the acyclic enamides into the corresponding cyclic enamides using the second generation Grubbs catalyst. With a limited number of substrates they observed a mixture of cyclic and acyclic products from the dimethyltitanocene reaction. Representative of their results was the generation of indole 2 in 40% overall yield from olefinic-amide 1. In addition to synthesizing indoles, they also generated dihydroquinolines and dihydroisoquinolines using this chemistry. More problematic was the use of the two-step protocol to generate sevenmembered ring substrates as olefin isomerization competed

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with cyclization. With some substrates they were able to overcome this problem by adding dihydroquinone to the reaction mixture.⁸

When combined with our experience with olefinic-ester and olefinic-lactone cyclizations, the Bennasar chemistry outlined above peaked our interest in studying olefinic-amide and olefinic-lactam cyclization reactions. Described here is the successful use of in situ generated titanium ethylidenes in this context.

To compare the titanium ethylidene reagent with Bennasar's two-step protocol, we decided to initially examine the cyclization of aromatic substrates 4–6. Concerned about the potential reaction of carbonyl protecting groups with the titanium reagent, i.e., Boc, CBz, etc., we opted to initially avoid this potential problem by employing a Ts group in this capacity. As illustrated in Table 1, the cyclization of

Table 1. Olefinic-Amide Cyclizations to Benzyl Fused Enamides

$$\begin{array}{c|c} & \text{TiCl}_4, \text{Zn, TMEDA} \\ & \text{PbCl}_2, \text{THF} \\ & \text{CH}_3\text{CHBr}_2 \\ & \text{C} & \text{Ts} \end{array} + \begin{array}{c|c} & \text{TiCl}_4, \text{Zn, TMEDA} \\ & \text{N} & \text{Ts} \\ & \text{Ts} & \text{A} & \text{Ts} \\ & \text{Ts} & \text{Ts}$$

entry	amide	n	enamide	yield $(\%)^a$	$C:A^b$
1	4	0	7	70	>95:5
2	5	1	8	78	>95:5
3	6	2	9	76	76:24

^a Combined yield of cyclic and acyclic enamides. ^b From isolated cyclic and acyclic enamide.

ene-amide **5** gave a 78% yield of dihydroquinoline **8C** (entry 2). Bennasar's two-step protocol on a related substrate resulted in a 41% overall yield. The cyclization to indole **7C** was also successful, giving it in 70% yield versus 40% overall using the two-step procedure (entry 1 and Scheme 1). Finally, the use of

Scheme 1. Bennasar Synthesis of Indole 2 via RCM

ene-amide 6 resulted in a 58% yield of seven-membered ring substrate 9C along with 18% of the corresponding acyclic

enamide **9A**. Bennasar's procedure was more competitive here; they were able to isolate a 50% overall yield of the seven-membered ring substrate along with 7% of the corresponding dihydroquinoline from olefin isomerization and cyclization. Olefin isomerization does not appear to be a problem with the titanium reagent.

Having established the viability of the olefinic-amide cyclizations to aromatic substrates, we next explored the cyclization of nonaromatic olefinic-lactams. When ε -caprolactam substrates 10, 11, and 12 were subjected to the titanium ethylidene reagent, we isolated cyclic enamides 14C, 15C, and 16C as the exclusive products (Table 2, entries

Table 2. Olefinic-Lactam Cyclizations

entry	amide	n	enamide	yield $(\%)^a$	$C:A^b$
1	10	1	14	70	>95:5 ^c
2	11	2	15	80	>95:5
3	12	3	16	82	>95:5
4	13	4	17	70	$86{:}14^d$

^a Combined yield of cyclic and acyclic enamide. ^b From isolated cyclic and acyclic enamides. ^c Characterized as the corresponding aminal and/or ketone (see Scheme 2). ^d Minor product was the corresponding lactam in which the alkene had undergone metathesis (see Supporting Information).

1–3). Noteworthy because it is unlikely that this substrate could be formed from an acyclic olefinic-enamide cyclization was the fact that this chemistry was successful in the generation of four-membered ring substrate 14C. ^{9,10} As with the aromatic substrate 6, lactam 13 gave a mixture of the cyclized seven-membered enamide 17C and an uncyclized product (entry 4).

Scheme 2. Enamide Derivatization

Not surprisingly enamide 14C was not stable to SiO₂ chromatography. For characterization purposes, we converted

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14C into cyclobutanone **18** by subjecting it to aqueous acid (Scheme 2).¹¹ Additionally, bromoaminal **19** was generated from the treatment of **14C** with NBS in MeOH.¹²

The lactam cyclizations are not limited to ε -caprolactam substrates. Cyclic enamides **22C** and **23C** were the major products from the cyclization of 2-pyrrolidinone **20** and δ -valerolactam **21**, respectively (entries 2 and 3, Table 3).

Table 3. Olefinic-Lactam Cyclizations

entry	amide	n	enamide	yield $(\%)^a$	$C:A^b$
1	12	3	16	82	>95:5
2	20	1	22	75	87:13
3	21	2	23	78	90:10

^a combined yield of cyclic and acyclic enamine. ^b ratio determined by ¹H NMR of the crude reaction mixture.

In contrast to the cyclization of ε -caprolactam substrate 12 to give 16C, however, these reactions did result in the formation of small amounts of acyclic enamide.

As a final test of the methodology, we examined the cyclization of methyl- and ethyl-substituted amides **24** and **25** (Scheme 3) When these substrates were subjected to the titanium ethylidene reagent, cyclic enamide was isolated in

Scheme 3. Olefinic-Amide Cyclizations

synthetically useful quantities but as a mixture with the corresponding acyclic enamides. As a mechanistic test, we did not observe the formation of **26C** when we resubjected acyclic enamide **26A** to the reaction conditions. This is consistent with the notion that cyclic enamide results from an olefin-metathesis, carbonyl-olefination pathway.¹³

In summary, we have demonstrated that titanium ethylidenes can be utilized in olefinic-amide and olefinic-lactam cyclization reactions and that the efficiency of the reaction depends somewhat on the nature of the substrate. We plan to continue examining the scope of olefinic-carbonyl cyclizations and their use in total synthesis efforts.

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Supporting Information Available: Experimental procedures and spectroscopic data for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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